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Simulation of Biochemical Reactions in Proteins

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Introduction



2007, Xiamen







2009, Heidelberg



2011, Singapore



2014, ECNU



Multiscale Simulations of Biochemical Systems





Outline

- Background of Quantum Mechanics
- Description of Simple Chemical Reactions
- Simulation of Chemical Reactions in Proteins



Motions of Micropaticles

Classical Mechanics:

Newton's second law:

F = ma

Quantum Mechanics:

Schrödinger equation:

$$\left(-\frac{h^2}{8\pi^2 m}\nabla^2 + \hat{V}\right)\psi = E\psi$$







Atomic Orbitals(AO) of Hydrogen Atom





Molecular Orbitals(MO) of Dihydrogen Molecule





Molecular Orbitals(MO) of H₂O





Center for Computational Chemistry at NYU Shanghai

Transition State Theory



Reaction Coordinate

 $\text{Reaction: HO}^{\text{-}} + \text{CH}_3\text{Br} \rightarrow [\text{HO}\text{---}\text{CH}_3\text{---}\text{Br}]^{\ddagger} \rightarrow \text{CH}_3\text{OH} + \text{Br}\text{--}$





Theoretical Pathways of Organometallic Reactions



Fei Xia's study



Transition State and IR Vibrations



Fei Xia's study



QM/MM Simulation of Chemical Reactions in Proteins



QM: quantum mechanics methods

MM: molecular mechanics methods



Function of Ras Protein



The QM/MM Simulation



Reaction Mechanism of GTP to GDP



Thank you for your attentions!

